

33. (a) The heat generated is the power output of the drill multiplied by the time: $Q = Pt$. We use $1 \text{ hp} = 2545 \text{ Btu/h}$ to convert the given value of the power to Btu/h and $1 \text{ min} = (1/60) \text{ h}$ to convert the given value of the time to hours. Then,

$$Q = \frac{(0.400 \text{ hp})(2545 \text{ Btu/h})(2.00 \text{ min})}{60 \text{ min/h}} = 33.9 \text{ Btu} .$$

- (b) We use $0.75Q = cm \Delta T$ to compute the rise in temperature. Here c is the specific heat of copper and m is the mass of the copper block. Table 19-3 gives $c = 386 \text{ J/kg}\cdot\text{K}$. We use $1 \text{ J} = 9.481 \times 10^{-4} \text{ Btu}$ and $1 \text{ kg} = 6.852 \times 10^{-2} \text{ slug}$ (see Appendix D) to show that

$$c = \frac{(386 \text{ J/kg}\cdot\text{K})(9.481 \times 10^{-4} \text{ Btu/J})}{6.852 \times 10^{-2} \text{ slug/kg}} = 5.341 \text{ Btu/slug}\cdot\text{K} .$$

The mass of the block is its weight W divided by the gravitational acceleration (which is 32 ft/s^2 in customary units, which uses “slugs” for mass):

$$m = \frac{W}{g} = \frac{1.60 \text{ lb}}{32 \text{ ft/s}^2} = 0.0500 \text{ slug} .$$

Thus,

$$\Delta T = \frac{0.750Q}{cm} = \frac{(0.750)(33.9 \text{ Btu})}{(5.341 \text{ Btu/slug}\cdot\text{K})(0.0500 \text{ slug})} = 95.3 \text{ K} = 95.3 \text{ C}^\circ .$$

This is equivalent to $(9/5)(95.3) = 172 \text{ F}^\circ$.