

32. Using  $v = f\lambda$  with  $v = 331 \text{ m/s}$  (see Table 18-1) with Eq. 20-2 and Eq. 20-25 leads to

$$f = \frac{v}{\left(\frac{1}{\sqrt{2}\pi d^2(N/V)}\right)} = (331 \text{ m/s})\pi\sqrt{2} (3.0 \times 10^{-10} \text{ m})^2 \left(\frac{nN_A}{V}\right) = \left(8.0 \times 10^7 \frac{\text{m}^3}{\text{s}\cdot\text{mol}}\right) \left(\frac{n}{V}\right)$$

Using the ideal gas law, we substitute  $n/V = p/RT$  into the above expression and find

$$f = \left(8.0 \times 10^7 \frac{\text{m}^3}{\text{s}\cdot\text{mol}}\right) \left(\frac{1.01 \times 10^5 \text{ Pa}}{(8.31 \frac{\text{J}}{\text{mol}\cdot\text{K}}) (273.15 \text{ K})}\right) = 3.5 \times 10^9 \text{ Hz} .$$

If we instead use  $v = 343 \text{ m/s}$  (the “default value” for speed of sound in air, used repeatedly in Ch. 18), then the answer is  $3.7 \times 10^9 \text{ Hz}$ .