5-1. (a) 
$$\lambda = \frac{h}{p} = \frac{h}{mv} = \frac{(6.63 \times 10^{-34} J \cdot s)(3.16 \times 10^7 s/y)}{(10^{-3} kg)(1 m/y)} = 2.1 \times 10^{-23} m$$

(b) 
$$v = \frac{h}{m\lambda} = \frac{6.63 \times 10^{-34} J \cdot s}{(10^{-3} kg)(10^{-2} m)} = 6.6 \times 10^{-29} m/s = 2.1 \times 10^{-21} m/y$$

5-2. 
$$\lambda = \frac{h}{p} \approx \frac{h}{E/c} = \frac{hc}{E} = \frac{1240 \, MeV \cdot fm}{100 \, MeV} = 12.4 \, fm$$

5-3. 
$$E_k = eV_o = \frac{p^2}{2m} = \frac{h^2}{2m\lambda^2} = \frac{(hc)^2}{2mc^2\lambda^2}$$
  $V_o = \frac{1}{e} \cdot \frac{(1240 \, eV \cdot nm)^2}{2(5.11 \times 10^5 \, eV)(0.04 \, nm)^2} = 940 \, V$ 

5-4. 
$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mE_k}} = \frac{hc}{\sqrt{2mc^2E_k}}$$
 (from Equation 5-2)

(a) For an electron: 
$$\lambda = \frac{1240 \, eV \cdot nm}{\left[ (2)(0.511 \times 10^6 \, eV)(4.5 \times 10^3 \, eV) \right]^{1/2}} = 0.0183 \, nm$$

(b) For a proton: 
$$\lambda = \frac{1240 \, eV \cdot nm}{\left[ (2)(938.3 \times 10^6 \, eV)(4.5 \times 10^3 \, eV) \right]^{1/2}} = 4.27 \times 10^{-4} \, nm$$

(c) For an alpha particle: 
$$\lambda = \frac{1240 \ eV \cdot nm}{\left[(2)(3.728 \times 10^9 \ eV)(4.5 \times 10^3 \ eV)\right]^{1/2}} = 2.14 \times 10^{-4} \ nm$$

5-5. 
$$\lambda = h/p = h/\sqrt{2mE_k} = hc/[2mc^2(1.5kT)]^{1/2}$$
 (from Equation 5-2)

Mass of  $N_2$  molecule =  $2 \times 14.0031 u(931.5 MeV/uc^2) = <math>2.609 \times 10^4 MeV/c^2 = 2.609 \times 10^{10} eV/c^2$ 

$$\lambda = \frac{1240 \, eV \cdot nm}{\left[ (2)(2.609 \times 10^{10} \, eV)(1.5)(8.617 \times 10^{-5} \, eV/K)(300 \, K) \right]^{1/2}} = 0.0276 \, nm$$

5-6. 
$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mE_k}} = \frac{hc}{\sqrt{2mc^2E_k}} = \frac{1240 \, eV \cdot nm}{\left[2(939.57 \times 10^6 \, eV)(0.02 \, eV)\right]^{\frac{1}{2}}} = 0.202 \, nm$$

- 5-7. (a) If there is a node at each wall, then  $n(\lambda/2) = L$  where n = 1, 2, 3, ... or  $\lambda = 2L/n$ .
  - (b)  $p = h/\lambda = hm/2L$   $E = p^2/2m = (hn/2L)^2/2m = h^2n^2/8mL^2$

$$E_n = \frac{(hc)^2 n^2}{8mc^2 L^2}$$

For n = 1: 
$$E_1 = \frac{(1240 \, eV \cdot nm)^2 (1)^2}{8 (938 \times 10^6 \, eV)(0.01 \, nm)^2} = 2.05 \, eV$$

For n = 2: 
$$E_2 = 2.05 \, eV(2)^2 = 8.20 \, eV$$

5-8. (a)  $\lambda/\lambda_c = 10^2$  is a nonrelativistic situation, so

$$\lambda/\lambda_c = [(hc/\sqrt{2mc^2E_k})/(hc/mc^2)] = (mc^2/2E_k)^{1/2}$$

$$E_k = \frac{mc^2}{2(\lambda/\lambda_c)^2} = \frac{0.511 \times 10^6 \, eV}{2(10^2)^2} = 25.6 \, eV$$

(b)  $\lambda/\lambda_c = 0.2$  is relativistic for an electron, so  $\lambda = h/\gamma mu \rightarrow \gamma u = h/\lambda m$ 

$$\frac{u/c}{\sqrt{1-(u/c)^2}} = \frac{h}{mc\lambda} = \frac{\lambda_c}{\lambda}$$

$$\frac{(u/c)^2}{1-(u/c)^2} = \left(\frac{\lambda_c}{\lambda}\right)^2 \rightarrow u/c = \frac{\lambda_c/\lambda}{\left[1+(\lambda_c/\lambda)^2\right]^{1/2}}$$

$$u/c = \frac{(1/0.2)}{[1 + (1/0.2)^2]^{1/2}} = 0.981 \rightarrow \gamma = 5.10$$

$$E_k = mc^2(\gamma - 1) = 0.511 MeV(\gamma - 1) = 2.10 MeV$$

(Problem 5-8 continued)

(c) 
$$\lambda/\lambda_c = 10^{-3}$$
 
$$u/c = \frac{(1/10^{-3})}{[1 + (1/10^{-3})^2]^{1/2}} = 0.9999 \rightarrow \gamma = 1000$$
 
$$E_k = mc^2(\gamma - 1) = 0.511 \, MeV(999) = 510 \, MeV$$

5-9. 
$$E_k = mc^2(\gamma - 1)$$
  $p = \gamma mu$   
(a)  $E_k = 2 \, GeV$   $mc^2 = 0.938 \, GeV$   
 $\gamma - 1 = E_k/mc^2 = 2 \, GeV/0.938 \, GeV = 2.132$  Thus,  $\gamma = 3.132$   
Because,  $\gamma = 1/\sqrt{1 - (u/c)^2}$  where  $u/c = 0.948$   

$$\lambda = \frac{h}{p} = \frac{h}{\gamma mc(u/c)} = \frac{hc}{\gamma mc^2(u/c)}$$

$$= \frac{1240 \, eV \cdot nm}{(3.132)(938 \times 10^6 \, eV)(0.948)} = 4.45 \times 10^{-7} \, nm = 0.445 \, fm$$
(b)  $E_k = 200 \, GeV$ 

$$E_k = 200 \text{ GeV}$$

$$\gamma - 1 = E_k / mc^2 = 200 \text{ GeV} / 0.938 \text{ GeV} = 213. \text{ Thus, } \gamma = 214 \text{ and } u/c = 0.9999$$

$$\lambda = \frac{1240 \text{ MeV: fm}}{(214)(938 \text{ MeV})(0.9999)} = 6.18 \times 10^{-3} \text{ fm}$$

5-10. 
$$n\lambda = D \sin \phi$$
 (Equation 5-5)  

$$\sin \phi = \frac{n\lambda}{D} = \frac{n}{D} \frac{hc}{\sqrt{2mc^2 E_k}} \text{ (see Problem 5-6)}$$

$$= \frac{1}{0.215 \, nm} \cdot \frac{1240 \, eV \cdot nm}{[2(5.11 \times 10^5 \, eV)]^{\frac{1}{2}} \sqrt{E_k}} = \frac{(5.705 \, eV)^{\frac{1}{2}}}{\sqrt{E_k}}$$

(Problem 5-10 continued)

(a) 
$$\sin \phi = \frac{(5.705 \, eV)^{\frac{1}{2}}}{\sqrt{75 \, eV}} = 0.659 \quad \phi = \sin^{-1}(0.659) = 41.2^{\circ}$$

(b) 
$$\sin \phi = \frac{(5.705 \, eV)^{\frac{1}{2}}}{\sqrt{100 \, eV}} = 0.570 \quad \phi = \sin^{-1}(0.570) = 34.8^{\circ}$$

5-11. 
$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2m_p E_k}} = 0.25 \, nm$$

Squaring and rearranging,

$$E_{k} = \frac{h^{2}}{2m_{p}\lambda^{2}} = \frac{(hc)^{2}}{2(m_{p}c^{2})\lambda^{2}} = \frac{(1240eV \cdot nm)^{2}}{2(938 \times 10^{6}eV)(0.25nm)^{2}}$$

$$E_{k} = 0.013eV$$

$$n\lambda = D\sin\phi \rightarrow \sin\phi = n\lambda/D = (1)(0.25nm)/0.304nm$$

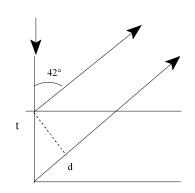
$$\sin\phi = 0.822 \rightarrow \phi = 55^{\circ}$$

5-12. (a) 
$$n\lambda = D \sin \phi$$
 :  $D = \frac{n\lambda}{\sin \phi} = \frac{nhc}{\sin \phi \sqrt{2mc^2 E_k}}$ 

$$= \frac{(1)(1240eV \cdot nm)}{(\sin 55.6^\circ)[2(5.11 \times 10^5 eV)(50eV)]^{\frac{1}{2}}} = 0.210 nm$$

(b) 
$$\sin \phi = \frac{n\lambda}{D} = \frac{(1)(1240 \, eV \cdot nm)}{(0.210 \, nm)[2(5.11 \times 10^5 \, eV)(100 \, eV)]^{\frac{1}{2}}}$$
  
= 0.584  $\phi = \sin^{-1}(0.584) = 35.7^{\circ}$ 

5-13.



$$d = t\cos 42^{\circ}$$

$$n\lambda = t + d = t(1 + \cos 42^{\circ}) = 0.30 \, nm(1 + \cos 42^{\circ})$$

For the first maximum n = 1, so  $\lambda = 0.523$  nm

$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mE_k}} \rightarrow E_k = \frac{h^2}{2m\lambda_2} = \frac{(hc)^2}{2mc^2\lambda^2}$$

$$E_k = \frac{(1240 \, eV \cdot nm)^2}{2(939.6 \times 10^6 \, eV)(0.523 \, nm)^2} = 3.0 \times 10^{-3} \, eV$$

5-14. 
$$\lambda = \frac{D\sin\phi}{n}$$
 (Equation 5-6)

For 54 eV electrons  $\lambda = 0.165 \text{ nm}$  and  $\sin \phi = (0.165 \text{ nm}) n / 0.215 \text{ nm} = 0.767 n$ 

For n = 2 and larger  $\sin \phi > 1$ , so no values of *n* larger than one are possible.

5-15. 
$$\sin \phi = n\lambda/D$$
 (Equation 5-6)

$$\lambda = h/p = h/\sqrt{2mE_k} = hc/\sqrt{2mc^2E_k}$$

$$= \frac{1240 \, eV \cdot nm}{\left[2(0.511 \times 10^6 \, eV)(350 \, eV)\right]^{1/2}} = 0.0656 \, nm$$

 $\sin \phi = n(0.0656 nm)/0.315 nm = 0.208 n$ 

For 
$$n=1$$
,  $\phi=12^{\circ}$  For  $n=2$ ,  $\phi=24.6^{\circ}$  For  $n=3$ ,  $\phi=38.6^{\circ}$ 

For n = 4,  $\phi = 56.4^{\circ}$  This is the largest possible  $\phi$ . All larger n values have  $\sin \phi > 1$ .

5-16. (a) 
$$\Delta t < \frac{1}{f} = \frac{1}{100.000 \, s^{-1}} = 10^{-5} \, s = 10 \, \mu s$$

(b) 
$$\Delta f \Delta t \approx \frac{1}{2\pi}$$
  $\therefore \Delta f \approx \frac{1}{2\pi \Delta t} = \frac{1}{2\pi \times 10^{-5} s} = 1.59 \times 10^4 \, Hz$ 

5-17. (a) 
$$y = y_1 + y_2$$
  

$$= 0.002 m \cos(8.0x/m - 400t/s) + 0.002 m \cos(7.6x/m - 380t/s)$$

$$= 2(0.002 m) \cos\left[\frac{1}{2}(8.0x/m - 7.6x/m) - \frac{1}{2}(400t/s - 380t/s)\right]$$

$$\times \cos\left[\frac{1}{2}(8.0x/m + 7.6x/m) - \frac{1}{2}(400t/s + 380t/s)\right]$$

$$= 0.004 m \cos(0.2x/m - 10t/s) \times \cos(7.8x/m - 390t/s)$$

(b) 
$$v = \frac{\overline{\omega}}{\overline{k}} = \frac{390/s}{7.8/m} = 50 \text{ m/s}$$

(c) 
$$v_g = \frac{\Delta \omega}{\Delta k} = \frac{20 / s}{0.4 / m} = 50 m / s$$

(d) Successive zeros of the envelope requires that  $0.2 \Delta x/m = \pi$ , thus  $\Delta x = \frac{\pi}{0.2} = 5 \pi m$  with

$$\Delta k = k_1 - k_2 = 0.4 m^{-1}$$
 and  $\Delta x = \frac{2\pi}{\Delta k}$ 

5-18. (a) 
$$v = f\lambda$$
 Thus,  $\frac{dv}{d\lambda} = f + \lambda \frac{df}{d\lambda}$ , multiplying by  $\lambda$ ,  $\lambda \frac{dv}{d\lambda} = \lambda f + \lambda^2 \frac{df}{d\lambda} = v + \frac{\lambda^2}{2\pi} \frac{d\omega}{d\lambda}$ 

$$-\frac{\lambda^2}{2\pi} \frac{d\omega}{d\lambda} = v - \lambda \frac{dv}{d\lambda} \text{ Because } k = 2\pi/\lambda, dk = -(2\pi/\lambda^2)d\lambda \text{ and}$$

$$\frac{d\omega}{dk} = v_g = v - \lambda \frac{dv}{d\lambda}$$

(b) v decreases as  $\lambda$  decreases.  $dv/d\lambda$  is positive.

5-19. (a) 
$$c = f\lambda = \lambda/T \rightarrow T = \lambda/c = 2 \times 10^{-2} m/3 \times 10^8 m/s = 6.7 \times 10^{-11} s/wave$$
  
The number of waves =  $0.25 \,\mu s/(6.7 \times 10^{-11} s/wave) = 3.73 \times 10^3$   
Length of the packet =  $(\# \text{ waves})(\lambda) = 2 \times 10^{-2} m (3.73 \times 10^3) = 74.6 m$ 

(b) 
$$f = c/\lambda = (3 \times 10^8 \, \text{m/s})/2 \times 10^{-2} \, \text{m} = 1.50 \times 10^{10} \, \text{Hz}$$

(c) 
$$\Delta\omega\Delta t \approx 1 \rightarrow \Delta\omega \approx 1/\Delta t = 1/0.25 \times 10^{-6} s = 4.0 \times 10^6 \ rad/s = 637 \ kHz$$

5-20. 
$$\Delta\omega\Delta t \approx 1 \rightarrow \Delta\omega \approx 1/\Delta t = 1/0.25 s = 4.0 \ rad/s \ or \ \Delta f \approx 0.6 \ Hz$$

5-21. 
$$\Delta\omega\Delta t\approx 1 \rightarrow (2\pi\Delta f)\Delta t=1$$
 Thus,  $\Delta t\approx 1/(2\pi\times5000)=3.2\times10^{-5}s$ 

5-22. (a) 
$$\lambda = h/p = h/\sqrt{2mE_k} = hc/\sqrt{2mc^2E_k}$$

$$= \frac{1240 \, eV \cdot nm}{[2(0.511 \times 10^6 \, eV) (5 \, eV)]^{1/2}} = 0.549 \, nm$$

 $d\sin\theta = \lambda/2$  For first minimum (see Figure 5-16).

$$d = \frac{\lambda}{2\sin\theta} = \frac{0.549 \, nm}{2\sin 5^{\circ}} = 3.15 \, nm \text{ slit separation}$$

(b) 
$$\sin 5^\circ = 0.5 \, cm/L$$
 where L = distance to detector plane  $L = \frac{0.5 \, cm}{2 \sin 5^\circ} = 5.74 \, cm$ 

- 5-23. (a) The particle is found with equal probability in any interval in a force-free region. Therefore, the probability of finding the particle in any interval  $\Delta x$  is proportional to  $\Delta x$ . Thus, the probability of finding the sphere *exactly* in the middle, i.e., with  $\Delta x = 0$  is zero.
  - (b) The probability of finding the sphere somewhere within 24.9cm to 25.1 cm is proportional to  $\Delta x = 0.2$  cm. Because there is a force free length L = 48 cm available to the sphere and the probability of finding it somewhere in L is unity, then the probability that it will be found in  $\Delta x = 0.2$  cm between 24.9 cm and 25.1 cm (or any interval of equal size) is:  $P\Delta x = (1/48)(0.2 cm) = 0.00417$ .

5-24. Because the particle must be in the box, 
$$\int_{0}^{L} \psi^* \psi dx = 1 = \int_{0}^{L} A^2 \sin^2(\pi x/L) dx = 1$$

Let  $u = \pi x/L$ ;  $x = 0 \rightarrow u = 0$ ;  $x = L \rightarrow u = \pi$  and  $dx = (L/\pi)du$ , so we have

$$\int_{0}^{\pi} A^{2}(L/\pi) \sin^{2}u \, du = A^{2}(L/\pi) \int_{0}^{\pi} \sin^{2}u \, du = 1$$

$$(L/\pi)A^2\int_0^\pi \sin^2 u \, du = (L/\pi)A^2\left[\frac{u}{2} - \frac{\sin 2u}{4}\right]_0^\pi = (L/\pi)A^2(\pi/2) = (LA^2)/2 = 1$$

$$A^2 = 2/L \rightarrow A = (2/L)^{1/2}$$

5-25. (a) At 
$$x = 0$$
:  $P dx = |\psi(0,0)|^2 dx = |A e^0|^2 dx = A^2 dx$ 

(b) At 
$$x = \sigma$$
:  $P dx = |A e^{-\sigma^2/4\sigma^2}|^2 dx = |A e^{-1/4}|^2 dx = 0.61 A^2 dx$ 

(c) At 
$$x = 2\sigma$$
:  $P dx = |Ae^{-4\sigma^2/4\sigma^2}|^2 dx = |Ae^{-1}|^2 dx = 0.14A^2 dx$ 

- (d) The electron will most likely be found at x = 0, where Pdx is largest.
- 5-26. (a) One does not know at which oscillation of small amplitude to start or stop counting.

$$f = \frac{N}{\Delta t}$$
  $\Delta f = \frac{\Delta N}{\Delta t} \approx \frac{1}{\Delta t}$ 

(b) 
$$\lambda = \frac{\Delta x}{N}$$
 and  $k = \frac{2\pi}{\lambda} = \frac{2\pi N}{\Delta x}$ , so  $\Delta k = \frac{2\pi \Delta n}{\Delta x} \approx \frac{2\pi}{\Delta x}$ 

5-27. 
$$\Delta E \Delta t \approx \hbar \rightarrow \Delta E \approx \hbar / \Delta t = \frac{1.055 \times 10^{-34} J \cdot s}{10^{-7} s (1.60 \times 10^{-19} J/eV)} \approx 6.6 \times 10^{-9} eV$$

5-28. 
$$\Delta x \, \Delta p \approx \hbar$$
  $\therefore \Delta x \approx \frac{\hbar}{\Delta p} = \frac{\hbar}{m \, \Delta v} = \frac{1.055 \times 10^{-34} J \cdot s}{(10^{-6} \times 10^{-3} \, kg)(0.01 \times 0.01 \, m/s)} \approx 10^{-21} m$ 

5-29. 
$$\Delta E \Delta t \approx \hbar \rightarrow \Delta E \approx \hbar / \Delta t = \frac{6.58 \times 10^{-16} \, eV \cdot s}{3.823 \, d_1 \cdot (8.64 \times 10^4 \, s/d_1)} \approx 1.99 \times 10^{-21} \, eV$$

The energy uncertainty of the excited state is  $\Delta E$ , so the  $\alpha$  energy can be no sharper than  $\Delta E$ .

5-30. 
$$\Delta x \Delta p \approx \hbar \rightarrow \lambda \Delta p \approx h \rightarrow \Delta p \approx h/\lambda$$
. Because  $\lambda - h/p$ ,  $p = h/\lambda$ ; thus  $\Delta p = p$ .

5-31. For the cheetah 
$$p = mv = 30 kg (40 m/s) = 1200 kg \cdot m/s$$
. Because  $\Delta p = p$  (see Problem 5-30),  $\Delta x \approx \hbar/\Delta p = 50 J \cdot s/1200 kg \cdot m/s \approx 4.2 \times 10^{-2} m = 4.2 cm$ 

5-32. Because 
$$c = f\lambda$$
 for photon,  $\lambda = cf = hc/hf = hc/E$ , so

$$E = hc/\lambda = \frac{1240 \, eV \cdot nm}{5.0 \times 10^{-3} nm} = 2.48 \times 10^5 \, eV$$

and 
$$p = E/c = \frac{2.48 \times 10^5 eV}{3.0 \times 10^8 m/s} = 8.3 \times 10^{-7} eV \cdot s/m$$

For electron:

$$\Delta p = h/\Delta x = \frac{4.14 \times 10^{-15} eV \cdot s}{5.0 \times 10^{-12} m} = 8.3 \times 10^{-4} eV \cdot s/m$$

Notice that  $\Delta p$  for the electron is 1000 times larger than  $\lambda$  for the photon.

5-33. (a) For <sup>48</sup> Ti: 
$$\Delta E_u(upper\ state) \approx \hbar/\Delta t = \frac{1.055 \times 10^{-34} J \cdot s}{1.4 \times 10^{-14} s \cdot (1.60 \times 10^{-13} J/MeV)} \approx 4.71 \times 10^{-10} MeV$$

$$\Delta E_L(lower\ state) \approx \hbar/\Delta t = \frac{1.055 \times 10^{-34} J \cdot s}{3.0 \times 10^{-12} s \cdot (1.60 \times 10^{-13} J/MeV)} \approx 2.20 \times 10^{-10} MeV$$

$$\Delta E (total) = \Delta E_u + \Delta E_L = 6.91 \times 10^{-10} MeV$$

$$\frac{\Delta E_T}{E} = \frac{6.91 \times 10^{-10} MeV}{1.312 MeV} = 5.3 \times 10^{-10}$$

(b) For Ha: 
$$\Delta E_u \approx \frac{1.055 \times 10^{-34} J \cdot s}{10^{-8} s (1.60 \times 10^{-19} J/eV)} \approx 6.59 \times 10^{-8} eV$$

and 
$$\Delta E_L \approx 6.59 \times 10^{-8} \, eV$$
 also.

 $\Delta E_T = 1.32 \times 10^{-7} \, eV$  is the uncertainty in the H $\alpha$  transition energy of 1.9 eV.

5-34. The size of the object needs to be of the order of the wavelength of the 10 MeV neutron.

 $\lambda = h/p = h/\gamma mu$ .  $\gamma$  and u are found from:

$$E_k = m_n c^2 (\gamma - 1)$$
 or  $\gamma - 1 = 10 MeV / 939 MeV$ 

$$\gamma = 1 + 10/939 = 1.0106 = 1/(1 - u^2/c^2)^{1/2}$$
 or  $u = 0.14 c$ 

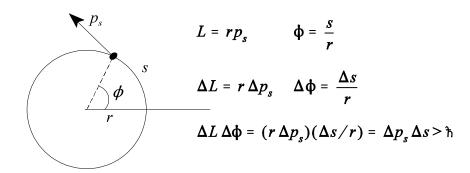
Then,  $\lambda = h/\gamma mu = hc/[\gamma mc^2(u/c)]$ 

$$= \frac{1240 \, eV \cdot nm}{[(1.0106)(939 \times 10^6 \, eV)(0.14)]} = 9.33 \times 10^{-6} \, nm = 9.33 \, fm$$

Nuclei are of this order of size and could be used to show the wave character of 10 MeV neutrons.

5-35. 
$$\Delta E \Delta t \approx \hbar$$
 :  $\tau \approx \Delta t \approx \frac{\hbar}{\Delta E} = \frac{1.055 \times 10^{-34} J \cdot s}{(1 eV)(1.602 \times 10^{-19} J/eV)} = 6.6 \times 10^{-16} s$ 

5-36.



In the Bohr model,  $L = n\hbar$  and may be known as well as  $\Delta L \approx 0.1 \,\hbar$ . Then  $\Delta \phi > \hbar/(0.1 \,\hbar) = 10 \, rad$ . This exceeds one revolution, so that  $\phi$  is completely unknown.

5-37. 
$$E = hf \rightarrow \Delta E = h\Delta f$$

$$\Delta E\Delta t \approx h \rightarrow \Delta f\Delta t \approx 1 \text{ where } \Delta t = 0.85 \text{ ms}$$

$$\Delta f = 1/0.85 \text{ ms} = 1.$$
For  $\lambda = 0.01 \text{ nm}$ 

$$f = c/\lambda = \frac{3.00 \times 10^8 \text{ m/s} \times 10^9 \text{ nm/s}}{0.01 \text{ nm}} ^{18 \times 10^9 \text{ Hz}}$$

$$\frac{\Delta f}{f} = \frac{1.18 \times 10^9 Hz}{3.00 \times 10^{19} Hz} = 3.9 \times 10^{-11}$$

5-38. 
$$\Delta p \Delta x \approx \hbar \rightarrow \Delta p \approx \hbar/\Delta x = \hbar/1 \text{ fm}$$
 Thus  $(\Delta p)^2 = \overline{p^2} = \hbar^2/(1 \text{ fm})^2$ 

For neutron:

 $f = 3.00 \times 10^{19} \text{ Hz}$ 

$$E = \frac{p^2}{2m_n} = \frac{\hbar^2}{(1fm)^2} \cdot \frac{1}{2m_n} = \frac{(\hbar c)^2}{2m_n c^2 (1fm)^2} = \frac{(197.3 \, MeV \cdot fm)^2}{2(939 \, MeV)(1fm)^2} = 20.7 \, MeV$$

For electron: The electron is relativistic, because classical kinetic energy  $p^2/2m$  is much larger than the rest energy. Therefore,

(Problem 5-38 continued)

$$E^{2} = (pc)^{2} + (m_{e}c^{2})^{2}$$

$$E^{2} = (\overline{h}c)^{2}/(fm)^{2} + (m_{e}c^{2})^{2}$$

$$E^{2} = (197.30 MeV \cdot fm)^{2}/(fm)^{2} + (0.511 MeV)^{2}$$

$$E^{2} = 3.8928 \times 10^{4} MeV^{2}$$

$$\therefore E = 197.30 MeV$$
and  $E_{\kappa} = 197.30 MeV - 0.511 MeV = 196.79 MeV$ 

5-39. (a) 
$$E^2 = p^2 c^2 + m^2 c^4$$
,  $E = hf = \hbar \omega$ ,  $p = h/\lambda = \hbar/k$ ,  $\hbar^2 \omega^2 = \hbar^2 k^2 c^2 + m^2 c^4$ 

$$v = \frac{\omega}{k} = \frac{\hbar \omega}{\hbar k} = \frac{\sqrt{\hbar^2 k^2 c^2 + m^2 c^4}}{\hbar k} = c\sqrt{1 + m^2 c^2/\hbar^2 k^2} > c$$

(b) 
$$v_g = \frac{d\omega}{dk} = \frac{d}{dk} \sqrt{\frac{k^2 c^2 + m^2 c^4}{\hbar k}}$$

$$= \frac{c^2 k}{\sqrt{\frac{k^2 c^2 + m^2 c^4}{\hbar^2}}} = \frac{c^2 k}{\omega} = \frac{c^2 \hbar k}{\hbar \omega} = \frac{c^2 p}{E} = u \qquad \text{(by Equation 2-41)}$$

5-40. 
$$\frac{\partial^2 y}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 y}{\partial t^2}$$
 (Equation 5-11)  $y_3 = c_1 y_1 + c_2 y_2$ 

$$\begin{split} \frac{\partial^2 y_3}{\partial x^2} &= C_1 \frac{\partial^2 y_1}{\partial x^2} + C_2 \frac{\partial^2 y_2}{\partial x^2} \\ &= C_1 \left( \frac{1}{v^2} \frac{\partial^2 y_1}{\partial t^2} \right) + C_2 1 \left( \frac{1}{v^2} \frac{\partial^2 y_2}{\partial t^2} \right) \\ &= \frac{1}{v^2} \frac{\partial^2}{\partial t^2} (C_1 y_1 + C_2 y_2) = \frac{1}{v^2} \frac{\partial^2 y_3}{\partial t^2} \end{split}$$

5-41. 
$$E = \overline{E} = \frac{\overline{p^2}}{2m} + \frac{1}{2}m\omega^2\overline{x^2} = \frac{(\Delta p)^2}{2m} + \frac{1}{2}m\omega^2(\Delta x)^2$$
. Substitute  $\Delta p = \frac{\hbar}{2\Delta x}$ 

$$E = \frac{\hbar^2}{8m(\Delta x)^2} + \frac{1}{2}m\omega^2(\Delta x)^2.$$
 To minimize E, set  $dE/d(\Delta x) = 0$ 

$$0 = \frac{dE}{d(\Delta x)} = \frac{\hbar^2}{8m} \frac{-2}{(\Delta x)^3} + m\omega^2 \Delta x = \frac{m\omega^2}{(\Delta x)^3} \left[ -\frac{\hbar^2}{4m^2\omega^2} + (\Delta x)^4 \right]$$

$$\therefore (\Delta x)^2 = \frac{\hbar}{2m\omega}$$

$$E_{\min} = \frac{\hbar^2}{8m} \cdot \frac{2m\omega}{\hbar} + \frac{m\omega^2}{2} \frac{\hbar}{2m\omega} = \frac{1}{4}\hbar\omega + \frac{1}{4}\hbar\omega = \frac{1}{2}\hbar\omega$$

$$\omega = \sqrt{\frac{K}{m}} = \sqrt{\frac{(1N/m)}{10^{-2}kg}} = 10/s$$

$$E_{\min} = \frac{1}{2} \hbar \omega = \frac{1}{2} (1.055 \times 10^{-34} J \cdot s)(10 / s) = 5.27 \times 10^{-34} J$$

5-42. (a) 
$$n(\lambda/2) = L \rightarrow \lambda = 2L/n$$
. Because  $\lambda = h/p = h/\sqrt{2mE}$ , then:

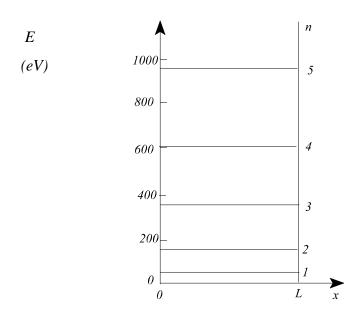
$$E = \frac{h^2}{2m\lambda^2} = \frac{h^2}{2m(2L/n)^2} = \frac{h^2n^2}{8mL^2}$$
 If  $E_1 = h^2/8mL^2$ , then:

$$E_n = \frac{h^2 n^2}{8mL^2} = n^2 E_1$$

(b) For L = 0.1 nm, 
$$E_1 = \frac{h^2}{8mL^2} = \frac{(hc)^2}{8mc^2L^2} = \frac{(1240\,eV\cdot nm)^2}{8(0.511\times 10^6 eV)(0.1\,nm)^2}$$

$$E_1 = 17.6 \, eV$$
 and  $E_n = 37.6 \, n^2 \, eV$ 

(Problem 5-42 continued)



(c) 
$$f = \Delta E/h \rightarrow c/\lambda = \Delta E/h \rightarrow \lambda = \frac{hc}{\Delta E}$$

For 
$$n = 2 \rightarrow n = 1$$
 transition,  $\Delta E = 112.8 \, eV$  and  $\lambda = \frac{1240 \, eV \cdot nm}{112.8 \, eV} = 11.0 \, nm$ 

(d) For 
$$n = 3 \rightarrow n = 2$$
 transition,  $\Delta E = 188 \, eV$  and  $\lambda = \frac{1240 \, eV \cdot nm}{188 \, eV} = 6.6 \, nm$ 

(e) For 
$$n = 5 \rightarrow n = 1$$
 transition,  $\Delta E = 903 \ eV$  and  $\lambda = \frac{1240 \ eV \cdot nm}{903 \ eV} = 1.4 \ nm$ 

5-43. (a) For proton: 
$$E_1 = \frac{(hc)^2}{8m_nc^2L^2}$$
 from Problem 5-42.

$$E_1 = \frac{(1240 \, MeV \cdot fm)^2}{8(938 \, MeV)(1 \, fm)^2} = 205 \, MeV \text{ and } E_n = 205 \, n^2 \, MeV$$

$$\therefore E_2 = 820 \, MeV \text{ and } E_3 = 1840 \, MeV$$

(Problem 5-43 continued)

(b) For  $n = 2 \rightarrow n = 1$  transition,

$$\lambda = \frac{hc}{\Delta E} = \frac{1240 \, MeV \cdot fm}{615 \, MeV} = 2.02 \, fm$$

(c) For  $n = 3 \rightarrow n = 2$  transition,

$$\lambda = \frac{1240 \, MeV \cdot fm}{1020 \, MeV} = 1.22 \, fm$$

(d) For  $n = 3 \rightarrow n = 1$  transition,

$$\lambda = \frac{1240 \, MeV \cdot fm}{1635 \, MeV} = 0.76 fm$$

- 5-44. (a)  $E \ge \hbar^2/2mL^2$  (Equation 5-31) And  $E = \hbar^2/2mA^2$ 
  - (b) For electron with  $A = 10^{-10} m$ :

$$E = \frac{(\hbar c)^2}{2mc^2A^2} = \frac{(197.3 \, eV \cdot nm)^2}{2(0.511 \times 10^6 \, eV) (10^{-1} \, nm)^2} = 3.81 \, eV$$

For electron with A = 1 cm or  $A = 10^{-2}$ 

$$E = 3.81 \, eV (10^{-1})^2 / (10^7 \, nm)^2 = 3.81 \times 10^{-16} \, eV$$

(c) 
$$E = \frac{\hbar^2}{2mL^2} = \frac{(1.055 \times 10^{-34} J \cdot s)^2}{2(100 \times 10^{-3} g \times 10^{-3} kg/g)(2 \times 10^{-2})^2} = 1.39 \times 10^{-61} J = 8.7 \times 10^{-43} eV$$

5-45. 
$$\Delta p = m \Delta v = m(0.0001)(500 \, m/s) = 0.05 \, m$$

For proton:  $\Delta x \Delta p \approx \hbar$ 

$$\Delta x \approx \hbar/\Delta p = (6.58 \times 10^{-16} \, eV \cdot s)/(0.05 \, m/s)(938 \times 10^6 \, eV)$$
  
$$\approx 1.40 \times 10^{-23} \, m = 1.40 \times 10^{-8} \, fm$$

For bullet: 
$$\Delta x \approx \frac{(1.055 \times 10^{-34} J \cdot s)}{(0.05 m/s)(10 \times 10^{-3} kg)} \approx 2.1 \times 10^{-31} m$$

5-46. 
$$\frac{\partial^2 y}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 y}{\partial t^2}$$
 (Equation 5-11) where  $y = f(\phi)$  and  $\phi = x - vt$ 

$$\frac{\partial y}{\partial x} = \frac{\partial f}{\partial \phi} \cdot \frac{\partial \phi}{\partial x} \quad \text{and} \quad \frac{\partial^2 y}{\partial x^2} = \frac{\partial f}{\partial \phi} \cdot \frac{\partial^2 \phi}{\partial x^2} + \frac{\partial \phi}{\partial x} \cdot \frac{\partial^2 f}{\partial \phi^2} \cdot \frac{\partial \phi}{\partial x}$$

$$\frac{\partial y}{\partial t} = \frac{\partial f}{\partial \phi} \cdot \frac{\partial \phi}{\partial t} \quad \text{and} \quad \frac{\partial^2 y}{\partial t^2} = \frac{\partial f}{\partial \phi} \cdot \frac{\partial^2 \phi}{\partial t^2} + \frac{\partial \phi}{\partial t} \cdot \frac{\partial^2 f}{\partial \phi^2} \cdot \frac{\partial \phi}{\partial t}$$

Noting that  $\frac{\partial^2 \phi}{\partial x^2} = 0$ ,  $\frac{\partial \phi}{\partial x} = 1$ ,  $\frac{\partial^2 \phi}{\partial t^2} = 0$ , and  $\frac{\partial \phi}{\partial t} = -v$ , we then have:

$$\frac{\partial f}{\partial \phi} \cdot 0 + 1 \cdot \frac{\partial^2 f}{\partial \phi^2} \cdot 1 = \frac{1}{v^2} \left( \frac{\partial f}{\partial \phi} \cdot 0 + (-v) \cdot \frac{\partial^2 f}{\partial \phi^2} \cdot (-v) \right)$$
$$\frac{\partial^2 f}{\partial \phi^2} = \frac{\partial^2 f}{\partial \phi^2}$$

5-47. (a)  $\lambda = h/p$  The electrons are not moving at relativistic speeds, so

$$\lambda = h/mv = 6.63 \times 10^{-34} J \cdot s / (9.11 \times 10^{-31} kg) (3 \times 10^6 m/s)$$
$$= 2.43 \times 10^{-10} m = 0.243 nm$$

(b) The energy, momentum, and wavelength of the two photons are equal.

$$E = \frac{1}{2}mv^{2} + mc^{2} = \frac{1}{2}mc^{2}(v^{2}/c^{2}) + mc^{2}$$

$$= mc^{2} \left[ \frac{1}{2}(v^{2}/c^{2}) + 1 \right]$$

$$= 0.511 \times 10^{6} \, eV \left[ \frac{1}{2}(3 \times 10^{6}/3 \times 10^{8})^{2} + 1 \right]$$

$$\approx 0.511 \, MeV$$

(c) 
$$p = E/c = 0.511 \, MeV/c$$

(d) 
$$\lambda = hc/E = 1240 \, eV \cdot nm/0.511 \times 10^6 \, eV = 2.43 \times 10^{-3} \, nm$$

5.48. (a) 
$$Q = m_p c^2 - m_n c^2 - m_\pi c^2$$
$$= 1.007825 u c^2 - 1.008665 u c^2 - 139.6 MeV$$
$$= 938.8 MeV - 939.6 MeV - 139.6 MeV$$
$$= -140.4 MeV$$
$$\Delta E = 140.4 MeV$$

(b) 
$$\Delta E \Delta t \approx \rightarrow \Delta t \approx \hbar/\Delta E = 6.58 \times 10^{-16} \, eV \cdot s / 140.4 \times 10^6 \, eV \approx 4.7 \times 10^{-24} s$$

(c) 
$$d = c \Delta t = 3.0 \times 10^8 \, m/s_1 (4.7 \times 10^{-24} \, s) = 1.4 \times 10^{-15} \, m = 1.4 \, fm$$

5-49. 
$$hf = \gamma mc^2 \rightarrow \gamma = \frac{hf}{mc^2} = \frac{1}{\sqrt{1 - v^2/c^2}}$$
$$1 - v^2/c^2 = \left(\frac{mc^2}{hf}\right)^2$$
$$\frac{v}{c} = \left[1 - \left(\frac{mc^2}{hf}\right)^2\right]^{1/2}$$

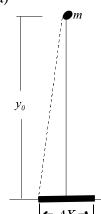
Expanding the right side, assuming  $mc^2 << hf$ ,

$$\frac{v}{c} = 1 - \frac{1}{2} \left( \frac{mc^2}{hf} \right)^2 - \frac{1}{8} \left( \frac{mc^2}{hf} \right)^4 + \cdots$$
 and neglecting all but the first two terms,

$$\frac{v}{c} = 1 - \frac{1}{2} \left( \frac{mc^2}{hf} \right)^2$$
 Solving this for m and inserting deBroglie's assumptions that

$$\frac{v}{c} \ge 0.99$$
 and  $\lambda = 30m$ , m is then:

$$m = \frac{[(1-0.99)2]^{1/2} (6.63 \times 10^{-34} J \cdot s)}{(3.00 \times 10^8 m/s)(30 m)} = 1.04 \times 10^{-44} kg$$



$$\Delta x \Delta p \approx \hbar \rightarrow m \Delta x \Delta v_x \approx \hbar \rightarrow \Delta v_x \approx \hbar / m \Delta x$$

$$y_0 = \frac{1}{2}gt^2 \rightarrow t = \left(\frac{2y_0}{g}\right)^{1/2} \qquad \frac{1}{2}\Delta X = \Delta v_x \cdot t = \Delta v_x \left(\frac{2y_0}{g}\right)^{1/2}$$

$$\Delta X = 2\Delta v_x \left(\frac{2y_0}{g}\right)^{1/2} = \frac{2\ln\left(\frac{2y_0}{g}\right)^{1/2}}{m\Delta x}$$

$$\Delta X = 2\Delta v_x \left(\frac{2y_0}{g}\right)^{1/2} = \frac{2\hbar \left(\frac{2y_0}{g}\right)^{1/2}}{m\Delta x}$$

(b) If also 
$$\Delta y \Delta p_y \approx \hbar \rightarrow \Delta v_y \approx \hbar/my$$
 and  $\frac{1}{2}\Delta X = \Delta x(t + \Delta t)$  where  $\Delta v_y = g\Delta t$  or

$$\Delta t = \Delta v_{v}/g = \hbar/mg\Delta y$$

so, 
$$\Delta X = \frac{2\hbar}{m\Delta x} [(2y_0/g)^{1/2} + \hbar/mg\Delta y]$$

5-51. 
$$\frac{1}{2}m\overline{v^2} = \frac{3}{2}kT$$

$$v_{rms} = \sqrt{\frac{3kT}{m}} = \left[\frac{3(1.381 \times 10^{-23} J/K)(300 K)}{56u(1.66 \times 10^{-27} kg/u)}\right]^{1/2} = 366 m/s$$

$$f' = f_o(1 + v/c) \rightarrow hf' = hf_o(1 + v/c)$$

$$\Delta E = hf' - hf_o = hf_o v/c = \frac{(1eV)(366 m/s)}{3.0 \times 10^8 m/s} = 1.2 \times 10^{-6} eV$$

This is about 12 times the natural line width.

$$\Delta E = h f_o v/c = \frac{(10^6 eV)(366 m/s)}{3.0 \times 10^8 m/s} = 1.2 eV$$

This is over  $10^7$  times the natural line width.

5-52. 
$$\rho_{recoil} = \rho_{\gamma} = E_{\gamma}/c \qquad E_{recoil} = \frac{(\rho_{recoil})^2}{2m} = \frac{E_{\gamma}^2}{2mc^2}$$

(a) 
$$E_{recoil} = \frac{(1eV)^2}{2(56uc^2)} \frac{uc^2}{931.5 \times 10^6 eV} = 9.6 \times 10^{-12} eV$$

This is about  $10^{-4}$  times the natural line width estimated at  $10^{-7}$  eV.

(b) 
$$E_{recoil} = \frac{(1 MeV)^2}{2(56 uc^2)} \frac{uc^2}{931.5 \times 10^6 eV} = 9.6 eV$$

This is about 10<sup>8</sup> times the natural line width.