

## Chapter 5

1. Starting with Equation (5.1), with  $n = 1$  and  $\theta = 15^\circ$ ,  $\sin \theta_1 = \frac{\lambda}{2d} = 0.259$

Second order:

$$\sin \theta_2 = \frac{2\lambda}{2d} = 2 \sin \theta_1 \quad \theta_2 = \sin^{-1}(2 \sin \theta_1) = 31.2^\circ$$

$$\theta_3 = \sin^{-1}(3 \sin \theta_1) = 50.9^\circ$$

- \* 2. Use  $\lambda = 0.186$  nm, and we know from the text that  $d = 0.282$  nm for NaCl.

$$n = 1: \quad \sin \theta = \frac{n\lambda}{2d} = \frac{\lambda}{2d} = 0.284 \quad \theta = 19.3^\circ$$

$$n = 2: \quad \sin \theta = \frac{n\lambda}{2d} = \frac{\lambda}{d} = 0.567 \quad \theta = 41.3^\circ$$

$$\Delta\lambda = 41.3^\circ - 19.3^\circ = 22.0^\circ$$

3. For  $n = 1$  we have  $\lambda = 2d \sin \theta = 2(0.314 \text{ nm})(\sin 12^\circ) = 0.131 \text{ nm}$

$$E = \frac{hc}{\lambda} = \frac{1240 \text{ eV} \cdot \text{nm}}{0.131 \text{ nm}} = 9.47 \text{ keV}$$

The largest order  $n$  is the largest integer for which  $\frac{n\lambda}{2d} < 1$

$$n < \frac{2d}{\lambda} = 4.79$$

so we can observe up through  $n = 4$ .

4. As in Davisson-Germer scattering  $n\lambda = D \sin \phi$

$$\phi = \sin^{-1}\left(\frac{\lambda}{D}\right) = \sin^{-1}\left(\frac{hc}{ED}\right) = \sin^{-1}\left(\frac{1240 \text{ eV} \cdot \text{nm}}{(10^5 \text{ eV})(0.24 \text{ nm})}\right) = 3.0^\circ$$

- 5.

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{(3.0 \text{ kg})(6.0 \text{ m/s})} = 3.68 \times 10^{-35} \text{ m}$$

No, the wavelength of the water waves depends on the medium; they are strictly mechanical waves.

6. Using the mean speed from kinetic theory (Chapter 9)

$$\bar{v} = \frac{4}{\sqrt{2\pi}} \sqrt{\frac{kT}{m}} = \frac{4}{\sqrt{2\pi}} \sqrt{\frac{(1.38 \times 10^{-23} \text{ J/K})(310.15 \text{ K})}{28(1.66 \times 10^{-27} \text{ kg})}} = 484.2 \text{ m/s}$$

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{28(1.66 \times 10^{-27} \text{ kg})(484.2 \text{ m/s})} = 2.94 \times 10^{-11} \text{ m}$$

or roughly 3% of the size of the molecule.

7. Using the approach of Example 5.2. and with  $K = e\Delta V$  we have, assuming relativistic effects are small,

$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mK}} = \frac{hc}{\sqrt{2(mc^2)K}}.$$

If we do not assume that relativistic effects are small, then

$$\lambda = \frac{h}{p} = \frac{hc}{\sqrt{K^2 + 2(mc^2)K}}.$$

See problem 11 for details. When  $K = 40 \text{ keV}$ , if we use the non-relativistic formula then

$$\lambda = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{2(0.511 \times 10^6 \text{ eV})(40 \times 10^3 \text{ eV})}} = 6.13 \times 10^{-3} \text{ nm} = 6.13 \text{ pm}.$$

Using the relativistic formula, we find

$$\lambda = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{(40 \times 10^3 \text{ eV})^2 + 2(0.511 \times 10^6 \text{ eV})(40 \times 10^3 \text{ eV})}} = 6.02 \text{ pm}.$$

This represents a 2% difference. Clearly when  $K = 100 \text{ keV}$ , we must use the relativistic approach and we find  $\lambda = 3.70 \times 10^{-3} \text{ nm} = 3.70 \text{ pm}$ .

- \* 8. The resolution will be comparable to the de Broglie wavelength. The energy of the microscope requires a relativistic treatment, so

$$\begin{aligned} \lambda &= \frac{h}{p} = \frac{hc}{\sqrt{K^2 + 2(mc^2)K}} \\ &= \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{(3 \times 10^6 \text{ eV})^2 + 2(0.511 \times 10^6 \text{ eV})(3 \times 10^6 \text{ eV})}} \\ &= 3.57 \times 10^{-4} \text{ nm} = 0.357 \text{ pm} \end{aligned}$$

$$9. (50 \text{ eV})(1.602 \times 10^{-19} \text{ J/eV}) = 8.01 \times 10^{-18} \text{ J}$$

$$\lambda = \frac{h}{\sqrt{2mK}} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{\sqrt{2(9.109 \times 10^{-31} \text{ kg})(8.01 \times 10^{-18} \text{ J})}} = 1.73 \times 10^{-10} \text{ m}$$

This is the same as in the textbook's example.

10. When  $E \gg E_0$  then  $E \approx pc$  for the particle;  $E = pc$  for a photon. Therefore the electron's energy is approximately equal to the photon energy.

If  $E = 2E_0$  then we cannot use  $E \approx pc$ . The exact expression is

$$p = \frac{\sqrt{E^2 - E_0^2}}{c} = \frac{\sqrt{3}E_0}{c}$$

for the electron's momentum. Then the de Broglie wavelength is

$$\lambda = \frac{h}{p} = \frac{hc}{\sqrt{3}E_0}$$

If the photon has the same wavelength, its energy is

$$E = \frac{hc}{\lambda} = \sqrt{3}E_0$$

\* 11. a) Relativistically

$$p = \frac{\sqrt{E^2 - E_0^2}}{c} = \frac{\sqrt{(K + mc^2)^2 - (mc^2)^2}}{c} = \frac{\sqrt{K^2 + 2Kmc^2}}{c}$$

$$\lambda = \frac{h}{p} = \frac{hc}{\sqrt{K^2 + 2Kmc^2}}$$

b) Non-relativistically, as in the text

$$\lambda = \frac{h}{\sqrt{2mK}} = \frac{hc}{\sqrt{2mc^2K}}$$

\* 12. The rest energy of the electron is very small compared to the total energy.

$$p = \frac{\sqrt{E^2 - E_0^2}}{c} = 50 \text{ GeV}/c$$

$$\lambda = \frac{h}{p} = \frac{1240 \text{ eV} \cdot \text{nm}}{50 \text{ GeV}} = 2.48 \times 10^{-17} \text{ m}$$

$$\text{fraction} = \frac{2.48 \times 10^{-17} \text{ m}}{2 \times 10^{-15} \text{ m}} = 0.012$$

13. a) For photons kinetic energy equals total energy and de Broglie wavelength is wavelength

$$K = E = \frac{hc}{\lambda} = \frac{1240 \text{ eV} \cdot \text{nm}}{0.15 \text{ nm}} = 8.27 \text{ keV}$$

b) The energy is low enough that we can use the non-relativistic formula:

$$K = \frac{h^2}{2m\lambda^2} = \frac{(hc)^2}{2mc^2\lambda^2} = \frac{(1240 \text{ eV} \cdot \text{nm})^2}{2(511 \times 10^3 \text{ eV})(0.15 \text{ nm})^2} = 66.9 \text{ eV}$$

c)

$$K = \frac{h^2}{2m\lambda^2} = \frac{(hc)^2}{2mc^2\lambda^2} = \frac{(1240 \text{ eV} \cdot \text{nm})^2}{2(939 \times 10^6 \text{ eV})(0.15 \text{ nm})^2} = 0.036 \text{ eV}$$

d)

$$K = \frac{h^2}{2m\lambda^2} = \frac{(hc)^2}{2mc^2\lambda^2} = \frac{(1240 \text{ eV} \cdot \text{nm})^2}{2(3727 \times 10^6 \text{ eV})(0.15 \text{ nm})^2} = 9.17 \times 10^{-3} \text{ eV}$$

14. a) As in Problem 6 we use the mean speed formula from kinetic theory:

$$\bar{v} = \frac{4}{\sqrt{2\pi}} \sqrt{\frac{kT}{m}} = \frac{4}{\sqrt{2\pi}} \sqrt{\frac{(1.38 \times 10^{-23} \text{ J/K})(5 \text{ K})}{(1.675 \times 10^{-27} \text{ kg})}} = 324 \text{ m/s}$$

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{(1.675 \times 10^{-27} \text{ kg})(324 \text{ m/s})} = 1.22 \text{ nm}$$

b)

$$\bar{v} = \frac{4}{\sqrt{2\pi}} \sqrt{\frac{kT}{m}} = \frac{4}{\sqrt{2\pi}} \sqrt{\frac{(1.38 \times 10^{-23} \text{ J/K})(0.01 \text{ K})}{(1.675 \times 10^{-27} \text{ kg})}} = 14.5 \text{ m/s}$$

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{(1.675 \times 10^{-27} \text{ kg})(14.5 \text{ m/s})} = 27.3 \text{ nm}$$

- \* 15. From the accelerating potential we know  $K = eV = 3 \text{ keV}$ .

$$E = K + E_0 = 514 \text{ keV}$$

$$p = \frac{\sqrt{E^2 - E_0^2}}{c} = \frac{\sqrt{(514 \text{ keV})^2 - (511 \text{ keV})^2}}{c} = 55.4 \text{ keV}/c$$

$$\lambda = \frac{h}{p} = \frac{hc}{pc} = \frac{1240 \text{ eV} \cdot \text{nm}}{55.4 \times 10^3 \text{ eV}} = 22.4 \text{ pm}$$

16. We use the relativistic formula derived in Problem 11:

$$\lambda = \frac{h}{p} = \frac{hc}{\sqrt{K^2 + 2Kmc^2}}$$

- a)  $\lambda = 0.194 \text{ nm}$       b)  $\lambda = 6.13 \times 10^{-2} \text{ nm}$       c)  $\lambda = 1.94 \times 10^{-2} \text{ nm}$   
d)  $\lambda = 6.02 \times 10^{-3} \text{ nm}$       e)  $\lambda = 1.64 \times 10^{-3} \text{ nm}$       f)  $\lambda = 2.77 \times 10^{-4} \text{ nm}$

From Example 5.1 in the text, we know the spacing of the NaCl lattice is  $0.282 \text{ nm}$ . So even the lowest energy electrons here could be used to probe the crystal structure.

17. a)

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{32 (1.661 \times 10^{-27} \text{ kg}) (480 \text{ m/s})} = 2.60 \times 10^{-11} \text{ m}$$

- b)

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{(1.5 \times 10^{-15} \text{ kg}) (10^{-6} \text{ m/s})} = 4.42 \times 10^{-13} \text{ m}$$

18. Using the relativistic formula from Problem 11

$$\lambda = \frac{h}{p} = \frac{hc}{\sqrt{K^2 + 2Kmc^2}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{(10^{12} \text{ eV})^2 + 2(10^{12} \text{ eV})(938 \times 10^6 \text{ eV})}} = 1.24 \times 10^{-18} \text{ m}$$

- 19.

$$d = D \sin(\phi/2) = (0.23 \text{ nm}) \sin 16^\circ = 0.063 \text{ nm}$$

$$\lambda = D \sin \phi = (0.23 \text{ nm}) \sin 32^\circ = 0.122 \text{ nm}$$

$$p = \frac{h}{\lambda} = \frac{hc}{\lambda c} = \frac{1240 \text{ eV} \cdot \text{nm}}{(0.122 \text{ nm}) c} = 10.2 \text{ keV}/c$$

$$E = \sqrt{p^2 c^2 + E_0^2} = \sqrt{(10.2 \text{ keV})^2 + (511 \text{ keV})^2} = 511.102 \text{ keV}$$

$$K = E - E_0 = 102 \text{ eV}$$

20. Beginning with Equation (5.7) and with  $V_0 = 48 \text{ V}$  and  $D = 0.215 \text{ nm}$  for nickel, we have

$$\lambda = \frac{1.226 \text{ nm} \cdot V^{1/2}}{\sqrt{48 \text{ V}}} = 0.177 \text{ nm}$$

$$\phi = \sin^{-1} \left( \frac{\lambda}{D} \right) = \sin^{-1} \left( \frac{0.177 \text{ nm}}{0.215 \text{ nm}} \right) = 55.4^\circ$$

At 64 eV

$$\lambda = \frac{1.226 \text{ nm} \cdot V^{1/2}}{\sqrt{64 V}} = 0.153 \text{ nm}$$

$$\phi = \sin^{-1} \left( \frac{\lambda}{D} \right) = \sin^{-1} \left( \frac{0.153 \text{ nm}}{0.215 \text{ nm}} \right) = 45.4^\circ$$

21. First we compute the wavelength of the electrons:

$$\lambda = \frac{h}{p} = \frac{hc}{\sqrt{E^2 - E_0^2}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{(513 \text{ keV})^2 - (511 \text{ keV})^2}} = 2.74 \times 10^{-2} \text{ nm}$$

From Figure 5.7(a) we see that  $2\theta_1 = \tan^{-1}(2.1 \text{ cm}/35 \text{ cm}) = 3.434^\circ$  or  $\theta_1 = 1.717^\circ$ . Now since  $\lambda = 2d \sin \theta$  we have

$$d_1 = \frac{\lambda}{2 \sin \theta_1} = \frac{2.74 \times 10^{-2} \text{ nm}}{2 \sin(1.717^\circ)} = 0.457 \text{ nm}$$

$$\theta_2 = \frac{1}{2} \tan^{-1}(2.3 \text{ cm}/35 \text{ cm}) = 1.880^\circ$$

$$d_2 = \frac{\lambda}{2 \sin \theta_2} = \frac{2.74 \times 10^{-2} \text{ nm}}{2 \sin(1.880^\circ)} = 0.412 \text{ nm}$$

$$\theta_3 = \frac{1}{2} \tan^{-1}(3.2 \text{ cm}/35 \text{ cm}) = 2.612^\circ$$

$$d_3 = \frac{\lambda}{2 \sin \theta_3} = \frac{2.74 \times 10^{-2} \text{ nm}}{2 \sin(2.612^\circ)} = 0.301 \text{ nm}$$

\* 22.

$$\lambda = \frac{h}{\sqrt{2mK}} = \frac{hc}{\sqrt{2mc^2K}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{2(939 \times 10^6 \text{ eV})(0.025 \text{ eV})}} = 0.181 \text{ nm}$$

$$\lambda = D \sin \phi \quad \phi = \sin^{-1} \left( \frac{\lambda}{D} \right) = \sin^{-1} \left( \frac{0.181 \text{ nm}}{0.45 \text{ nm}} \right) = 23.7^\circ$$

23. We begin with Equation (5.23)  $\Delta\omega \Delta t = 2\pi$ . Since  $\omega = 2\pi f$  the first equation is equivalent to  $\Delta f \Delta t = 1$ . Then  $\Delta t = \frac{1}{\Delta f} = \frac{1}{0.3 \text{ Hz}} = 3.33 \text{ s}$ . If we want the time to be one-half of that found from the bandwidth relation, then we must monitor the frequency of the system every 1.67 s.

24. Refer to problem 11. The 5 eV electrons are nonrelativistic. We must treat the 500 keV electrons relativistically.

For the 5 eV electrons,  $\lambda = \frac{hc}{\sqrt{2mc^2K}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{2(0.511 \times 10^6 \text{ eV})(5 \text{ eV})}} = 0.549 \text{ nm}$ . The electron speed can be found using  $p = mv = \frac{h}{\lambda}$ . Therefore

$$v = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{(9.11 \times 10^{-31} \text{ kg})(0.549 \times 10^{-9} \text{ m/s})} = 1.33 \times 10^6 \text{ m/s}$$

and  $\beta = \frac{v}{c} = 4.42 \times 10^{-3}$ . Alternatively for this nonrelativistic case you could use  $K = 5 \text{ eV} = \frac{1}{2}mv^2$  to find  $v$ .

The phase velocity is found using a modification of Equation (5.32). We must note that  $E$  represents the total energy of the particle, so

$$v_{ph} = \frac{\omega}{k} = \frac{E/\hbar}{p/\hbar} = \frac{E}{p} = \frac{0.511 \times 10^6 \text{ eV} (1.602 \times 10^{-19} \text{ J/eV})}{(9.11 \times 10^{-31} \text{ kg}) (1.33 \times 10^6 \text{ m/s})} = 6.76 \times 10^{10} \text{ m/s}.$$

This speed exceeds the speed of light but is not associated with the transmission of information. In problem 28 we will show that a simpler approach to find the phase velocity is to use  $v_{ph} = \frac{c}{\beta}$  but either approach yields the same answer. We find the group velocity from

Equation (5.31), namely  $u_{gr} = \frac{pc^2}{E}$ . In problem 73 from chapter 2, we showed that  $\beta = \frac{pc}{E}$  so  $u_{gr} = \beta c = 1.33 \times 10^6 \text{ m/s}$  which is the same as the particle speed.

For the 500 keV electrons,

$$\lambda = \frac{hc}{\sqrt{K^2 + 2(mc^2)K}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{(500 \times 10^3 \text{ eV})^2 + 2(0.511 \times 10^6 \text{ eV})(500 \times 10^3 \text{ eV})}}$$

so  $\lambda = 1.42 \times 10^{-3} \text{ nm}$ . Using the same approach as before  $p = \frac{h}{\lambda} = 4.67 \times 10^{-22} \text{ kg} \cdot \text{m/s}$ .

However, this is a relativistic momentum so  $p = \gamma m v$ . We find  $\gamma = \frac{K + E_0}{E_0} = 1.979$  and

therefore  $v = 2.59 \times 10^8 \text{ m/s} = 0.863 c$ . The phase velocity is  $v_{ph} = \frac{c}{\beta} = 1.16 c$  and again the group velocity is the same as the particle velocity,  $u_{gr} = 0.863 c$ .

25. a)  $f = \frac{v}{\lambda} = 0.571 \text{ Hz}$

b) From the initial conditions given, we should use a cosine function.

$$\Psi = A \cos \left( \left( \frac{2\pi}{\lambda} \right) (x - vt) \right) = (3.0 \text{ cm}) \cos \left( \left( \frac{2\pi}{7 \text{ cm}} \right) (10 \text{ cm} - (4 \text{ cm/s})(13 \text{ s})) \right) = 3.0 \text{ cm}$$

26. a)  $f = \frac{v}{\lambda} = \frac{4.2 \text{ cm/s}}{4.0 \text{ cm}} = 1.05 \text{ Hz}$

b)  $T = 1/f = 0.95 \text{ s}$

c)  $k = 2\pi/\lambda = \pi/2 \text{ cm}^{-1}$

d)  $\omega = 2\pi/T = 2.1\pi \text{ rad/s}$

27. a)

$$\Psi = \Psi_1 + \Psi_2 = 0.003 [\sin(6.0x - 300t) + \sin(7.0x - 250t)]$$

We can use a trig identity  $\sin A + \sin B = 2 \sin \left( \frac{A+B}{2} \right) \cos \left( \frac{A-B}{2} \right)$

$$\Psi = 0.006 \sin(6.5x - 275t) \cos(-0.5x - 25t)$$

or because cosine is an even function

$$\Psi = 0.006 \sin(6.5x - 275t) \cos(0.5x + 25t)$$

At 64 eV

$$\lambda = \frac{1.226 \text{ nm} \cdot V^{1/2}}{\sqrt{64 \text{ V}}} = 0.153 \text{ nm}$$

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24. Refer to problem 11. The 5 eV electrons are nonrelativistic. We must treat the 500 keV electrons relativistically.

$$\text{For the 5 eV electrons, } \lambda = \frac{hc}{\sqrt{2mc^2K}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{2(0.511 \times 10^6 \text{ eV})(5 \text{ eV})}} = 0.549 \text{ nm. The electron}$$

speed can be found using  $p = mv = \frac{h}{\lambda}$ . Therefore

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b)  $T = 1/f = 0.95 \text{ s}$

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b)

$$v_{ph} = \frac{\omega_1 + \omega_2}{k_1 + k_2} = \frac{550 \text{ rad/s}}{13 \text{ m}^{-1}} = 42.3 \text{ m/s}$$

$$u_{gr} = \frac{\Delta\omega}{\Delta k} = \frac{50 \text{ rad/s}}{1 \text{ m}^{-1}} = 50 \text{ m/s}$$

c) As in Equation (5.22)  $\Delta x = 2\pi/\Delta k = 2\pi \text{ m}$  is the separation between zeros.d)  $\Delta k \Delta x = (1 \text{ m}^{-1})(2\pi \text{ m}) = 2\pi$ 

28.

$$u_{gr} = \frac{d\omega}{dk} = \frac{dE}{dp} = \frac{d}{dp} (p^2 c^2 + E_0^2)^{1/2} = \frac{pc^2}{\sqrt{p^2 c^2 + E_0^2}} = \frac{pc^2}{E} = \beta c$$

$$v_{ph} = \lambda v = \frac{h \omega}{p} = \frac{E}{p} = \frac{pc^2/v}{p} = \frac{c^2}{v} = \frac{c}{\beta}$$

The particle and its “signal” are associated with the group velocity, not the phase velocity.

- \* 29. As in Example 5.6 we start with  $v_{ph} = c\lambda^n$  where  $c$  is a constant. We also know from Equation (5.33) that  $u_g = v_{ph} + k \frac{dv_{ph}}{dk}$ . We note further that  $\frac{dv_{ph}}{dk} = \frac{dv_{ph}}{d\lambda} \left( \frac{d\lambda}{dk} \right)$ . Since  $\lambda = \frac{2\pi}{k}$  then  $\frac{d\lambda}{dk} = \frac{-2\pi}{k^2} = \frac{-\lambda^2}{2\pi}$ . Therefore

$$u_g = v_{ph} + \left( \frac{2\pi}{\lambda} \right) \frac{dv_{ph}}{d\lambda} \left( \frac{-\lambda^2}{2\pi} \right) = v_{ph} - (\lambda) \frac{dv_{ph}}{d\lambda}$$

Setting  $u_g = v_{ph} = c\lambda^n$ , we find  $c\lambda^n = c\lambda^n - cn\lambda^n$ . This can be satisfied only if  $n = 0$ , so  $v_{ph}$  is independent of  $\lambda$ . This is consistent with the idea that when a medium is non-dispersive, the phase and group velocities are equal and the speed independent of wavelength.

30. Protons:

$$\gamma = \frac{K + E_0}{E_0} = \frac{946.27 \text{ MeV}}{938.27 \text{ MeV}} = 1.00853 \quad \beta = \sqrt{1 - \frac{1}{1.00853^2}} = 0.130$$

$$u_{gr} = \beta c = 0.130c \quad v_{ph} = \frac{c}{\beta} = 7.7c$$

Electrons:

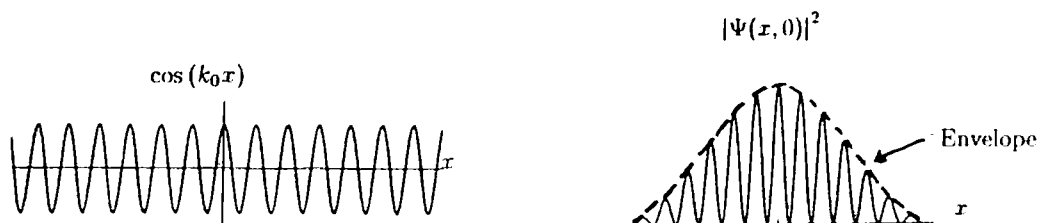
$$\gamma = \frac{K + E_0}{E_0} = \frac{8.511 \text{ MeV}}{0.511 \text{ MeV}} = 16.66 \quad \beta = \sqrt{1 - \frac{1}{16.66^2}} = 0.9982$$

$$u_{gr} = \beta c = 0.9982c \quad v_{ph} = \frac{c}{\beta} = 1.002c$$

31.

$$\begin{aligned} \Psi(x, 0) &= \int \tilde{A}(k) \cos(kx) dk = A_0 \int_{k_0 - \Delta k/2}^{k_0 + \Delta k/2} \cos(kx) dk \\ &= A_0 \frac{\sin(kx)}{x} \Big|_{k_0 - \Delta k/2}^{k_0 + \Delta k/2} = \frac{A_0}{x} (\sin(k_0 + \Delta k/2)x - \sin(k_0 - \Delta k/2)x) \\ &= \frac{2A_0}{x} \sin\left(\frac{\Delta kx}{2}\right) \cos(k_0 x) \end{aligned}$$

See the diagrams below. At the half-width of  $|\Psi(x, 0)|^2$ , we have  $\sin^2\left(\frac{\Delta k x}{2}\right) = \frac{1}{2}$ , so  $\frac{\Delta k x}{2} = \frac{\pi}{4}$  and  $x = \frac{\pi}{(2\Delta k)}$ . Then  $\Delta x = 2x = \frac{\pi}{\Delta k}$ , and  $\Delta k \Delta x = \pi$ .



\* 32. Relativistically

$$u = \frac{dE}{dp} = \frac{d}{dp} (p^2 c^2 + E_0^2)^{1/2} = \frac{pc^2}{\sqrt{p^2 c^2 + E_0^2}} = \frac{pc^2}{E}$$

Classically

$$u = \frac{dE}{dp} = \frac{d}{dp} \left( \frac{p^2}{2m} \right) = \frac{p}{m}$$

33. For a double slit, the amplitude of  $E$  is doubled, and hence the intensity (proportional to  $E^2$ ) is higher by a factor of four for the double slit.

34.

$$d = \frac{\lambda}{\sin \theta} = \frac{h/p}{\sin \theta} = \frac{hc}{(\sqrt{E^2 - E_0^2}) \sin \theta} = \frac{1240 \text{ eV} \cdot \text{nm}}{(\sqrt{(512 \text{ keV})^2 - (511 \text{ keV})^2}) \sin 1^\circ} = 2.22 \text{ nm}$$

\* 35. We make use of the small angle approximations:  $\sin \theta \approx \tan \theta$  and  $\sin \theta \approx \theta$ .

$$\sin \theta \approx \tan \theta = \frac{0.3 \text{ mm}}{0.8 \text{ m}} = 3.75 \times 10^{-4}$$

$$\lambda = d \sin \theta \approx d \theta = (2000 \text{ nm}) (3.75 \times 10^{-4}) = 0.75 \text{ nm}$$

$$p = \frac{h}{\lambda} = \frac{hc}{\lambda c} = \frac{1240 \text{ eV} \cdot \text{nm}}{(0.75 \text{ nm}) c} = 1.653 \text{ keV}/c$$

$$K = E - E_0 = \sqrt{p^2 c^2 + E_0^2} - E_0 = \sqrt{(1.653 \text{ keV})^2 + (511 \text{ keV})^2} - 511 \text{ keV} = 2.67 \text{ eV}$$

Such low energies will present problems, because low-energy electrons are more easily deflected by stray electric fields.

36. By the uncertainty principle  $\Delta p \Delta x = \frac{\hbar}{2}$  at minimum. Non-relativistically (with  $\Delta x = d$ )

$$E_{\min} = \frac{(\Delta p)^2}{2m} = \frac{\hbar^2}{8md^2} = \frac{(\hbar c)^2}{8mc^2 d^2} = \frac{(197.3 \text{ eV} \cdot \text{nm})^2}{8(938.27 \times 10^6 \text{ eV})(1.6 \times 10^{-5} \text{ nm})^2} = 20.3 \text{ keV}$$

37. For the  $n = 1$  energy level

$$E = \frac{\hbar^2}{8md^2} = \frac{\hbar^2 c^2}{8mc^2 d^2} = \frac{(1240 \text{ eV} \cdot \text{nm})^2}{8(939.57 \times 10^6 \text{ eV})(2 \times 10^{-6} \text{ nm})^2} = 51.1 \text{ MeV}$$

By the uncertainty principle  $\Delta p \Delta x = \frac{\hbar}{2}$  at minimum. Non-relativistically (with  $\Delta x = d$ )

$$E_{\min} = \frac{(\Delta p)^2}{2m} = \frac{\hbar^2}{8md^2} = 1.30 \text{ MeV}$$

The two answers differ by the factor of  $(2\pi)^2$  in using  $\hbar$  or  $h$  in the energy formula.

- \* 38. The uncertainty ratio is the same for any mass and independent of the box length.

$$\Delta p \Delta x = m \Delta v \Delta x \geq \frac{\hbar}{2} \quad \text{or} \quad \Delta v \geq \frac{\hbar}{2m \Delta x} = \frac{\hbar}{2mL}$$

$$E = \frac{1}{2}mv^2 = \frac{\hbar^2}{8mL^2} \quad \text{or} \quad v = \sqrt{\frac{\hbar^2}{4m^2 L^2}} = \frac{\hbar}{2mL}$$

$$\frac{\Delta v}{v} = \frac{\hbar/2mL}{\hbar/2mL} = \frac{1}{2\pi}$$

39. For circular motion  $L = rp$  and so  $\Delta L = r \Delta p$ . Along the circle  $x = r\theta$  and  $\Delta x = r \Delta \theta$ . Thus

$$\Delta p \Delta x = \frac{\Delta L}{r} (r \Delta \theta) = \Delta L \Delta \theta \geq \frac{\hbar}{2}$$

For complete uncertainty  $\Delta \theta = 2\pi$  and

$$\Delta L = \frac{\hbar/2}{2\pi} = \frac{\hbar}{4\pi}$$

40.  $\Delta E \Delta t \geq \frac{\hbar}{2}$  so

$$\Delta E \geq \frac{\hbar}{2\Delta t} = \frac{1.054 \times 10^{-34} \text{ J} \cdot \text{s}}{(2 \times 10^{36} \text{ y})(3.16 \times 10^7 \text{ s/y})} = 1.67 \times 10^{-78} \text{ J}$$

41. If we use Equation (5.44),  $\Delta \omega \Delta t = \frac{1}{2}$  we find

$$\Delta \omega = \frac{1}{2\Delta t} = \frac{1}{2(2.4 \mu\text{s})} = 2.1 \times 10^5 \text{ Hz}$$

$$42. \Delta p \Delta x = m \Delta v \Delta x \geq \frac{h}{2} \text{ so at minimum uncertainty}$$

$$\Delta v = \frac{2m\Delta x}{h} = \frac{2(3 \times 10^{-15} \text{ kg})(10^{-6} \text{ m})}{1.054 \times 10^{-34} \text{ J} \cdot \text{s}} = 1.76 \times 10^{-14} \text{ m/s}$$

$$* 43. \text{ a) } \Delta E \Delta t \geq \frac{h}{2} \text{ so}$$

$$\Delta E \geq \frac{h}{2\Delta t} = \frac{6.582 \times 10^{-16} \text{ eV} \cdot \text{s}}{2(1 \times 10^{-13} \text{ s})} = 3.29 \times 10^{-3} \text{ eV}$$

b) Using the photon relation  $E = \frac{hc}{\lambda}$  and taking a derivative

$$dE = -\frac{hc}{\lambda^2} d\lambda = -\frac{hc}{E^2} d\lambda$$

Then letting  $\Delta\lambda = d\lambda$  and  $\Delta E = dE$  we have

$$|\Delta\lambda| = hc \frac{\Delta E}{E^2} = (1240 \text{ eV} \cdot \text{nm}) \frac{(4.7 \text{ eV})}{3.29 \times 10^{-3} \text{ eV}^2} = 0.185 \text{ nm}$$

\* 44. The wavelength of the electrons should be 0.1 nm or less. For this wavelength

$$p = \frac{h}{\lambda} = \frac{hc}{1240 \text{ eV} \cdot \text{nm}} = \frac{\lambda c}{(0.14 \text{ nm})} = 8.86 \text{ keV}/c$$

$$K = E - E_0 = \sqrt{p^2 c^2 + E_0^2} - E_0 = \sqrt{(8.86 \text{ keV})^2 + (511 \text{ keV})^2} - 511 \text{ keV} = 77 \text{ eV}$$

45. For the angle  $\theta_R$  we find  $\theta_R \approx \tan \theta_R = \frac{400 \text{ nm}}{20 \text{ cm}} = 2 \times 10^{-5}$ . The wavelength is

$$\lambda = \frac{d \theta_R}{1.22} = \frac{(0.05 \text{ m})(2 \times 10^{-5})}{1.22} = 820 \text{ nm}$$

a)

$$E = \frac{hc}{\lambda} = \frac{1240 \text{ eV} \cdot \text{nm}}{820 \text{ nm}} = 1.51 \text{ eV}$$

b) For non-relativistic electrons

$$K = \frac{p^2}{2m} = \frac{h^2}{2m\lambda^2} = \frac{2mc^2\lambda^2}{(1240 \text{ eV} \cdot \text{nm})^2} = \frac{2(511 \times 10^3 \text{ eV})(820 \text{ nm})^2}{2} = 2.24 \times 10^{-6} \text{ eV}$$

Check with the uncertainty principle:

$$\Delta p \geq \frac{h}{2\Delta x} = \frac{1.055 \times 10^{-34} \text{ J} \cdot \text{s}}{2(4000 \times 10^{-9} \text{ m})} = 1.32 \times 10^{-29} \text{ kg} \cdot \text{m/s}$$

The actual momentum is

$$p = \frac{h}{\lambda} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{820 \times 10^{-9} \text{ m}} = 8.08 \times 10^{-28} \text{ kg} \cdot \text{m/s}$$

This is allowed because  $p > \Delta p$ .

46.

$$\lambda = \frac{h}{p} = \frac{h}{\sqrt{2mK}} = \frac{1240 \text{ eV} \cdot \text{nm}}{\sqrt{2(3727 \text{ MeV})(5.5 \text{ MeV})}} = 6.12 \text{ fm}$$

The minimum kinetic energy according to the uncertainty principle is

$$K = \frac{(\Delta p)^2}{2m} = \frac{(\hbar c)^2}{8mc^2 (\Delta x)^2} = \frac{(197.33 \text{ eV} \cdot \text{nm})^2}{8(3727 \text{ MeV})(16 \times 10^{-6} \text{ nm})^2} = 5.10 \text{ keV}$$

Since the kinetic energy exceeds the minimum, it is allowed.

\* 47. The proof is done in Example 6.11. With  $\omega = \sqrt{k/m}$  we have a minimum energy

$$E = \frac{\hbar\omega}{2} = \frac{\hbar}{2} \sqrt{\frac{k}{m}} = \frac{1.055 \times 10^{-34} \text{ J} \cdot \text{s}}{2} \sqrt{\frac{8.2 \text{ N/m}}{0.0023 \text{ kg}}} = 3.15 \times 10^{-33} \text{ J}$$

48. We can determine the value using the normalization condition:

$$\int_0^L A^2 \sin^2\left(\frac{\pi x}{L}\right) dx = 1 = A^2 \frac{L}{2} \quad \text{or} \quad A = \sqrt{\frac{2}{L}}$$

No answers are provided for questions 49 through 53.

54.

$$\psi_1 = A \sin\left(\frac{\pi x}{L}\right) \quad \psi_2 = A \sin\left(\frac{2\pi x}{L}\right) \quad \psi_3 = A \sin\left(\frac{3\pi x}{L}\right)$$

where  $A = \sqrt{2/L}$ . Refer to problem 48.

55. As before  $\lambda = \frac{2L}{n}$  so  $p_n = \frac{nh}{2L}$ . At high energies we must use relativity, so

$$E = \sqrt{p^2 c^2 + E_0^2} = E_0 \sqrt{\frac{p^2 c^2}{E_0^2} + 1}$$

$$\frac{E_2}{E_1} = \left[ \frac{1 + h^2 c^2 / (L^2 E_0^2)}{1 + h^2 c^2 / (4L^2 E_0^2)} \right]^{1/2}$$

$$\frac{E_3}{E_1} = \left[ \frac{1 + 9h^2 c^2 / (4L^2 E_0^2)}{1 + h^2 c^2 / (4L^2 E_0^2)} \right]^{1/2}$$

$$\frac{E_4}{E_1} = \left[ \frac{1 + 4h^2 c^2 / (L^2 E_0^2)}{1 + h^2 c^2 / (4L^2 E_0^2)} \right]^{1/2}$$

These are quite different from the non-relativistic results, as one might expect. They do reduce to the non-relativistic results in the low-energy limit.

56. At time  $t = 0$  the velocity is uncertain by at least  $\Delta v_0 = \frac{\Delta p}{m} = \frac{\hbar}{2m\Delta x}$ . After a time  $t = \frac{m(\Delta x)^2}{\hbar}$  we have

$$\Delta x' = (\Delta v_0)t = \frac{\hbar}{2m\Delta x} \frac{m(\Delta x)^2}{\hbar} = \frac{\Delta x}{2}$$

which means the uncertainty equals half the distance of travel.

57. Both the spatial distribution  $\psi(x)$  and the wavenumber distribution  $\phi(k)$  should have the same Gaussian form:

$$\psi(x) \propto \exp\left[-\frac{x^2}{(2\Delta x)^2}\right] \quad \phi(k) \propto \exp\left[-\frac{k^2}{(2\Delta k)^2}\right]$$

For conjugate variables  $(x, k)$  it is possible to obtain one distribution by taking a Fourier transform of the other. Letting  $A$  be a normalization constant for  $\phi(k)$  we have

$$\psi(x) = \frac{A}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dk \exp\left[-\frac{k^2}{(2\Delta k)^2}\right] \exp(ikx)$$

The integral is done by completing the square:

$$\begin{aligned} \psi(x) &= \frac{A}{\sqrt{2\pi}} \int_{-\infty}^{\infty} dk \exp\left[-\frac{k^2}{4\Delta k^2} + ikx - x^2\Delta k^2 + x^2\Delta k^2\right] \\ &= \frac{A}{\sqrt{2\pi}} \exp(-x^2\Delta k^2) \int_{-\infty}^{\infty} dk \exp\left[\left(-\frac{1}{4\Delta k^2}\right)(k - 2ix\Delta k^2)^2\right] \end{aligned}$$

Letting  $u = \frac{(k - 2ix\Delta k^2)}{2\Delta k}$  we have

$$\psi(x) = A\sqrt{\frac{2}{\pi}} \Delta k \exp(-x^2\Delta k^2) \int_{-\infty}^{\infty} \exp(-u^2) du$$

The integral has a value  $\sqrt{\pi}$  so

$$\psi(x) = \sqrt{2}A\Delta k \exp(-x^2\Delta k^2)$$

Now comparing with the Gaussian form

$$\psi(x) \propto \exp\left[-\frac{x^2}{(2\Delta x)^2}\right]$$

we see that

$$(\Delta k)^2 = \frac{1}{(2\Delta x)^2} \quad \text{or} \quad \Delta k \Delta x = \frac{1}{2}$$

58.  $\Delta E \Delta t \geq \hbar/2$

$$\Delta E \geq \frac{\hbar}{2\Delta t} = \frac{6.582 \times 10^{-16} \text{ eV} \cdot \text{s}}{2 \times 10^{-16} \text{ s}} = 3.29 \text{ eV}$$

\* 59.

$$\Delta t = \frac{d}{c} = \frac{1.2 \times 10^{-15} \text{ m}}{3.0 \times 10^8 \text{ m/s}} = 4.0 \times 10^{-24} \text{ s}$$

$$\Delta E \geq \frac{\hbar}{2\Delta t} = \frac{6.582 \times 10^{-16} \text{ eV} \cdot \text{s}}{2(4.0 \times 10^{-24} \text{ s})} = 82 \text{ MeV}$$

This “lower bound” estimate of the rest mass is  $\Delta E/c^2$  which is within a factor of two of the rest energy.

60. a) In general  $n\lambda = 2d \sin \theta$  so

$$\sin \theta = \frac{n\lambda}{2d} = n \frac{0.5 \text{ nm}}{2(0.8 \text{ nm})} = 0.3125 n$$

It is required that  $\sin \theta \leq 1$  so the allowed values of  $n$  are 1, 2, 3. Substituting we find  $\theta = 18.2^\circ$  for  $n = 1$ ,  $\theta = 38.7^\circ$  for  $n = 2$ , and  $\theta = 69.6^\circ$  for  $n = 3$ .

b) For electrons

$$p = \frac{h}{\lambda} = \frac{hc}{\lambda c} = \frac{1240 \text{ eV} \cdot \text{nm}}{(0.5 \text{ nm}) c} = 2480 \text{ eV}/c$$

$$K = E - E_0 = \sqrt{p^2 c^2 + E_0^2} - E_0 = \sqrt{(2.480 \text{ keV})^2 + (511 \text{ keV})^2} - 511 \text{ keV} = 6.0 \text{ eV}$$

61. The uncertainty of the strike zone width is  $\Delta x < 0.38 \text{ m}$ . There is a second uncertainty, in the  $x$ -component of the ball's velocity, given by

$$\Delta p_x = m \Delta v_x \geq \frac{\hbar}{2 \Delta x}$$

Due to the uncertainty in velocity the ball's  $x$  position will be uncertain by the time it reaches home plate in an amount

$$\Delta v_x t = \frac{\hbar t}{2m \Delta x} \leq 0.38 \text{ m}$$

where we have added the inequality because this is the condition for a strike. Since this inequality must be satisfied simultaneously with the first one  $\Delta x < 0.38 \text{ m}$ , we multiply the two inequalities together to find

$$\frac{\hbar t}{2m} \leq (0.38 \text{ m})^2 = 0.144 \text{ m}^2$$

Rearranging we find

$$\hbar \leq \frac{2m (0.144 \text{ m}^2)}{t} = \frac{2 (0.145 \text{ kg}) (0.144 \text{ m}^2)}{(18 \text{ m}) / (35 \text{ m/s})} = 0.081 \text{ J} \cdot \text{s}$$

62. Assume that the given angle corresponds to the first order reflection. We have:

$$\lambda = 2d \sin \theta = 2 (0.156 \text{ nm}) \sin 26^\circ = 0.1368 \text{ nm}.$$

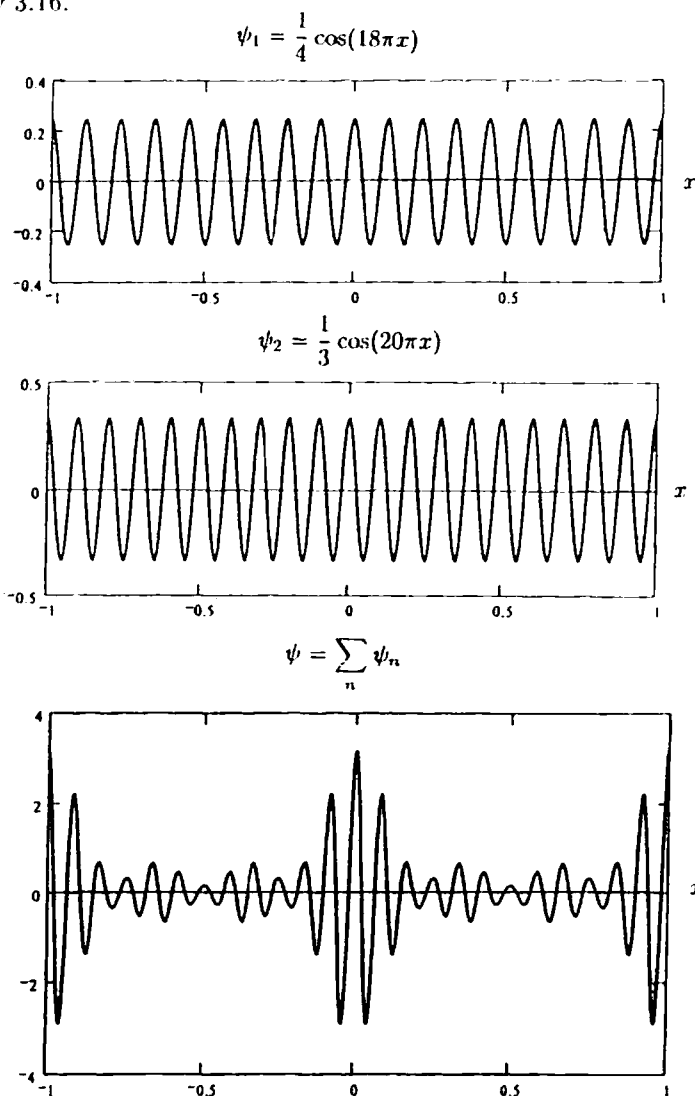
Next we find the energy:

$$p = \frac{h}{\lambda} = \frac{hc}{\lambda c} = \frac{1240 \text{ eV} \cdot \text{nm}}{(0.1368 \text{ nm}) c} = 9.06 \times 10^3 \text{ eV}/c$$

$$\begin{aligned} K &= E - E_0 = \sqrt{p^2 c^2 + E_0^2} - E_0 = \sqrt{(9.06 \text{ keV})^2 + (939.57 \text{ MeV})^2} - 939.57 \text{ MeV} \\ &= 4.36 \times 10^{-2} \text{ eV} \end{aligned}$$

The Oak Ridge Electron Linear Accelerator Pulsed Neutron Source (ORELA) produces intense, nanosecond bursts of neutrons, each burst containing neutrons with energies from  $10^{-3}$  to  $10^8 \text{ eV}$ .

63. a) Using the given information, the wave packet will equal:  $\psi = \sum_n \psi_n = \frac{1}{4} \cos(18\pi x) + \frac{1}{3} \cos(20\pi x) + \frac{1}{2} \cos(22\pi x) + \cos(24\pi x) + \frac{1}{2} \cos(26\pi x) + \frac{1}{3} \cos(28\pi x) + \frac{1}{4} \cos(30\pi x)$ . A graph of the first two terms and a graph of the entire packet are shown below.  $\psi_9$  has a frequency of 9 and an amplitude of 0.25.  $\psi_{10}$  has a frequency of 10 and an amplitude of 0.33. Other terms would have a frequency of 11 and amplitude of 0.5, etc. The peak value of the packet is approximately 3.16.



- b) The packet is centered about  $x = 0$  but extends to  $\pm\infty$ . The packet repeats every unit along the  $x$  axis.

- \* 64. a) Starting from Equation (5.45),  $\Delta E \Delta t \geq \frac{\hbar}{2}$  and substituting we have  $\left(\frac{\Gamma}{2}\right) \tau \geq \frac{\hbar}{2}$ . Therefore  $\Gamma \tau \geq \hbar$ .

- b) We can find the minimum value for  $\Gamma$  from the equation above. Using the data for the



neutron, for example,

$$\Gamma_{\text{neutron}} = \frac{6.582 \times 10^{-16} \text{ eV} \cdot \text{s}}{887 \text{ s}} = 7.42 \times 10^{-19} \text{ eV}.$$

The other values follow in a similar fashion.

$$\Gamma_{\text{pion}} = 2.53 \times 10^{-8} \text{ eV} \qquad \Gamma_{\text{upsilon}} = 65.8 \text{ keV}$$